Synthesis of thiadiazine, imidazothiadiazole, diazospiroundecatetraene and spirothiadiazolopyrimidinocyclohexadiene derivatives from 2,5-dithiobiureas

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Ethenetetracarbonitrile (2, in tetrahydrofuran solution), diethyl (*E*)-2,3-dicyanobutenedioate (10, in tetrahydrofuran solution), and 7,7',8,8'-tetracyanoquinodimethane (16, in pyridine solution) act on 1-substituted-2,5-dithiobiureas 1a–c, forming the derivatives of imidazothiadiazole 5a–c, thiadiazine 11a–c and 12a–c, spiro(thiadiazolopyrimidino-cyclohexadiene)malononitrile 17a–c and diazospiroundecatetraene 18a–c. Rationales for these conversions involving the nucleophilic addition on diycanomethylene carbon atom are presented.

Keywords: ethenetetracarbonitrile, diethyl (*E*)-2,3-dicyanobutenedioate and tetracyanoquinodimethane, heterocyclisation of substituted 2,5-dithiobiureas

Ethenetetracarbonitrile (tetracyanoethylene, **2**) is the simplest of percyanoalkenes (cyanocarbons).¹ It shows a great affinity for electrons, and is thus a fairly good dehydrogenating agent towards dihydroheteroaromatic systems.¹ The general chemistry of **2** had been reviewed with emphasis on molecular complexes by an intermolecular charge-transfer interaction,^{1.4} tricyanovinylation reactions,¹ additions¹ and cycloaddition reactions⁵ as well as cyanation reagents.⁶⁻⁸ Ethenetetracarbonitrile (**2**) reacted with thiosemicarbazides to give a variety of heterocyclic compounds *via* a single-electron transfer mechanism.⁹⁻¹⁶

Diethyl (*E*)-2,3-dicyanobutenedioate have been discovered to be a viable and versatile alternatives to tetracyanoethylene as building blocks for the synthesis of molecule-based magnets.¹⁷ The reaction of diethyl (*E*)-2,3-dicyanobutenedioate (**10**) with ethylenediamine, *trans*-1,2-diaminocyclohexane afforded piprazinone and quinoxalinone derivatives, respectively.¹⁸ Also, diethyl (*E*)-2,3-dicyanobutenedioate reacted with *o*-phenylenediamine and 2,3-diaminonaphthalene with the formation of quinoxaline and quinoxalinone derivatives.¹⁹ Aminopteridine derivatives and related compounds have been prepared during the reaction of compound **10** with uracil and thiouracil derivatives.²⁰

Recently, we have demonstrated that 1,6-disubstituted-2,5dithiobiureas 1 reacted with 2 to give products 3 and 4, in addition to disubstituted-1,3,4-thiadiazoles (Fig. 1).¹³



Fig. 1

As well be outlined in detail below, in this paper we reported several heterocyclisation of 1-substituted-2,5-dithiobiureas 1a-c using ethenetetracarbonitrile (2) either as a reaction mediator or as a building block, diethyl (*E*)-2,3-dicyanobutenedioate (10) and 7,7',8,8'-tetracyanoquinodimethane (16) as π -acceptors.



Scheme 1

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Results and discussion

Addition of doubled molar amounts of 2 to solutions of 1a-c in tetrahydrofuran (THF) with admission of air resulted in a pink colouration of the solution which quickly turned into brown. This behaviour may be explained as due to initial formation of unstable charge-transfer (CT) complexes followed by chemical reaction. Monitoring of the reaction by visible spectroscopy failed since the reaction is fast and also at lower concentration no significant colour changes were observed. After concentration of the reaction mixture, the residue was subjected to vacuum sublimation to remove any unreacted TCNE. Chromatographic separations of the residue gave numerous coloured zones. From the two most intense zones, products **5a–c** and **6a–c** could be isolated (Scheme 1).

The structures of **5a–c** are supported by characteristic IR, ¹H and ¹³C NMR as well as mass spectral data by the following findings: The gross formula of **5a**, $C_{12}H_6N_6S$ represents a product from one molecule of **1a** and one molecule of **2** with loss of (H₂S + 2HCN). The IR spectrum of **5a** shows absorption characteristic of NH, CN and C=N groups at 3380, 2215 and 1635 cm⁻¹, respectively. The low-field NH attached to phenyl group is present at $\delta_H = 9.76$ ppm. In its ¹³C NMR spectrum, C-2, C-5, C-6 and C-4a resonate at $\delta_C = 156.88$, 122.24, 122.26 and 154.53 ppm, respectively. Further peaks at 118.11 and 118.14 ppm (CN), besides the aromatic carbons support the assigned structure.

The formations of 9 did not take place in the absence of 2 hence its presence is definitely required for the cyclisation observed. Charge-transfer complexes may (but not necessarily) play an intermediate role. Since the cyclisation involves intramolecular nucleophilic attacks on either of the thiocarbonyl groups. It is conceivable that 2 accelerates the process through intermediates 7 and 8 (Scheme 2), activating the respective C=S bonds toward nucleophilic addition. After cyclisation and formation of intermediates 9, 2 is released with liberation of H₂S.

Two routes could be suggested, the first one is the tendency of **2** to effect tetracyanoethylation the intermediate **9** followed by the elimination two molecules of HCN and formation the imidazothiadiazole derivatives **5a–c**. The second route is the formation of the intermediate **9**' to give the product **6a–c** (Scheme 2). Thus, ethenetetracarbonitrile (**2**) may act either as a mediator or as a building block in heterocyclisation of compounds **1a–c**. The ¹H NMR spectrum of **6b** showed two broad signals with the ratio 1:1 centred at 6.57 and 8.87 ppm, due to hydrazine-NH and phenyl-NH protons. The signal around 4.61 in ¹H NMR and 52.58 ppm in ¹³C NMR spectra due to benzylic-CH₂ further supports the structure assigned to **6b**. The ¹³C-DEPT NMR spectrum exhibiting negative signal at $\delta = 52.58$ ppm. The decoupled carbon spectrum of **6b** showed signals at $\delta = 155.48$ and 154.21 ppm due to thiazole-C-2 and C-5, respectively, besides the aromatic carbons. The IR spectrum of **6b** showed absorption bands at 3385–3360, 1625 and 1600 due to NH, C=N and (ArC=C) groups.

The molecular ions of **6a–c** in their EI-mass spectra confirm the molecular masses and the gross compositions. Furthermore, the following common features of the fragmentation patterns support the assigned structure: Loss of N₂ giving intense (M⁺-28) ions and loss of RNCS giving rise to the ion m/z = 261 common in the spectra of all three compounds. The odd electron fragments represented the three isothiocyanates indicating that the charge during fragmentation may remain with the isothiocyanate fragments.

We now report a convenient method for one-step synthesis of thiadiazine **11a–c** and **12a–c** derivatives. This method consists of the reaction of **1a–c** with **10** (in a 1:2 molar ratio). The weaker acceptor **10**, was expected to react similarly to **2** with the donors **1a–c**. The reaction was carried out in THF at room temperature and nearly two days were required to achieve conversion of 96% for **1a–c** and medium yields of **11** and **12** were observed (Scheme 3).

Recombination of 1a-c ion radical under abstraction of two hydrogen protons by 10° to give the dimer A along with 10-H^{\circ}. Elimination of a molecule of S₂ from the dimer A afforded the radical 13° . Combination of the radicals 13° and 10-H^{\circ} gave the adduct 14. The resulting adduct 14 undergoes elimination of one molecule of ethyl formate followed by intramolecular nucleophilic attack to the ethylenic double bond with the elimination of HCN to form the thiadiazine derivatives 11a-c.

On the other hand, the thiadiazines **12a–c** were obtained from cyclisation of the adduct **14** *via* an elimination a molecule of ethyl cyanoacetate (Scheme 4).

The IR spectrum of **11b** showed absorption bands due to NH₂ and NH stretching vibration at 3350 and 3225 cm⁻¹, an α , β -unsaturated nitrile CN at 2225, carbonyl ester at 1685 and C=N group at 1620 cm⁻¹. ¹H NMR spectrum of **11b** in DMSO-



Scheme 2



Scheme 3

d₆, revealed signals at 4.62 due to benzylic-CH₂, 1.25 and 4.16 for ethoxy group (J = 7.10), two broad signals with the ratio 2:1 centred at 6.34 and 8.89 ppm, due to exocyclic NH₂ and benzylic-NH, respectively. Signals around 52.64 (CH₂), 118.19 (CN), 155.92, 161.44 (C-5, C-2) and 168.81 (COO) in the ¹³C NMR spectrum further support structure **11b**. The molecular formula of compounds **11a–c** were obtained and confirmed by elemental analysis and mass data. The presence of ethoxy and benzylic groups are also evident from the ¹³C-DEPT NMR spectrum exhibiting positive signals at 14.26 and negative signals at 52.64 and 60.79 ppm.

For the formation of thiadiazines 12a-c, the IR spectrum of 12c shows characteristic bands at v = 3360, 3270 (NH₂ and NH), 2220 (CN) and 1685 (COO) as well as C=N at 1610

cm⁻¹. The ¹H NMR spectrum of **12c** shows two broad signals at $\delta_{\rm H} = 6.35$ and 7.55 ppm, due to NH₂ as well as allyl-NH. Also, the ¹H NMR spectrum of **12c** shows one singlet and two multiplets at $\delta_{\rm H} = 4.09$, 5.18–5.26 and 5.90–5.94 ppm, due to allyl group. The decoupled carbon spectra of **12c** showed signals at $\delta_{\rm C} = 14.19$ (CH₃), 42.88 (allyl-CH₂N), 60.63 (CH₂O), 116.06 (allyl-CH₂=), 118.16 (CN), and 134.22 ppm, (allyl-CH=). Also, the ¹³C NMR of **12c** shows another signals at $\delta_{\rm C} = 158.93$, 159.67 and 166.12 ppm, due to (C-2, C-5) and (COO), respectively. The gross formula $C_{10}H_{13}N_5O_2S$ represents a product from one molecule of **1c** and one of **10** with loss of sulfur and ethyl cyanoacetate.

7,7',8,8'-Tetracyanoquinodimethane (TCNQ, **16**) is well known electron acceptor molecule,²¹ which has been successfully used for the preparation of electrically conducting salts and CT-complexes.²² The interest of TCNQ has focused on its potential applications on material rectifiers,²³ non-linear optical materials,²⁴ organoferromagnets²⁵ and organic chromophores.²⁶

Recently, it has been reported that the addition of 1,8diaminonaphthalene to TCNQ (**16**) afforded 2-(4-(1*H*,3*H*pyrimidin-2-ylidene)cyclohexa-2,5-dienylidene)malononitrile.²⁷ Pyridine solution of TCNQ (**16**) and **1a–c** in a molar ratio of 2:1 were kept at 100°C for 2 h with admission of air. Chromatographic separation of the residue after concentration gave numerous coloured zones, from which products **17** and **18** could be isolated (Scheme 5).

Structural assignment of products 17a-c is based on spectral data and on combustion analysis. The elemental analysis of 17b supporting the gross formula $C_{21}H_{14}N_8S$, and the mass spectrum which gave a correct molecular ion at m/z = 410 (23%).



Scheme 4



Scheme 5

The IR spectrum of **17b** showed NH₂ and NH absorption bands at v = 3395 and 3315, cyano groups at 2220 and 2215 cm⁻¹. The ¹H NMR spectrum revealed two broad signals at $\delta = 6.70$ and 8.91 ppm, related to NH₂ and benzyl-NH, respectively, besides the aromatic and cyclohexadiene protons. The ¹³C NMR spectrum of **17b** confirmed its ¹H NMR spectral data by appearance of signals at 4.62, 6.30 and 6.58 due to benzylic-CH₂ and cyclohexadiene-CH. The ¹³C NMR spectrum of **17b** revealed signals for pyrimidine (C-6, C-5) which resonate at $\delta = 60.45$ and 160.91 ppm, respectively are accordance with the observed trends in the δ values for C-atoms in push-pull alkenes.^{28,29}

The second zone characterised by a brown colour contains compounds spiropyridazine-cyclohexadiene derivatives **18a–c**. The IR spectrum of **18a** showed characteristic absorption to the NH₂ and NH groups at 3385, 3270, 1570 (NH def. and C-N str.) and 1358 as well as 998 cm⁻¹, to strong vibrational coupling of C=S and C-N entities. The ¹H NMR spectrum of **18a** showed three broad signals with the ratio 2:2:1 centred at 6.49, 6.62 and 9.84 due to two exocyclic-NH₂ and Ph-NH protons. The decoupled carbon spectrum of **18a** showed signals at $\delta = 180.19$ (C=S), 60.71 (C-5), 156.67 (C-3), 160.83 (C-6) and 46.21 (q-C-4,1'). The analytical data of compound **18a** could also match for other isomers of products **19** and **20** (Fig. 2).

The alternative structures **19** and **20** could be ruled out on the bases of ¹H NMR, ¹³C NMR and the fragment ions in the mass spectrum of **18a** at m/z 263, 207, 197, 135, 66 and 57.

As shown in Fig. 2, structure **18a** fits best to all the spectroscopic data (see experimental section).



Scheme 6

Formation of spiro compounds **17** and **18** derivatives from TCNQ may be rationalised as outlined in (Scheme 6).

Experimental

Melting points have been determined using open capillaries on a Gallenkamp melting point apparatus and are uncorrected. Elemental analysis were determined by Microanalytical Centre, Cairo University, Egypt. The IR spetra were recorded with Shimadzue 408 instrument using potassium bromide. 500 MHz ¹H and 125 MHz ¹³C



NMR spectra on a Bruker DRX 500 spectrometer. Chemical shifts are expressed as δ [ppm] with reference to tetramethylsilane as an internal standard, s = singlet, d = doublet, dd = doublet of doublet, t = triplet, m = multiplet and br = broad. ¹³C assignments (q C = sp² quaternary carbon atoms) were made with the aid of distortionless enhancement by polarisation transfer (DEPT) 135/90 spectra. Mass spectra were recorded on Varian MAT 311 instrument in EI Mode (70 eV) ionisation energy. Preparative layer chromatography (PLC) used air-dried 1.0 mm thick layers of slurry applied silica gel, Merck Pf₂₅₄ on 48 cm wide and 20 cm high glass plates using the solvents listed. Zones were to 254 nm light and eluted with acetone.

Starting materials

1-Substituted-2,5-dithiobiureas $1\mathbf{a}$ -c were prepared according to published procedures, as were 1-phenyl-2,5-dithiobiurea $(1\mathbf{a})$,³⁰ 1benzyl-2,5-dithiobiurea $(1\mathbf{b})$,^{30,31} 1-allyl-2,5-dithiobiurea $(1\mathbf{c})$.³⁰ Ethenetetracarbonitrile (2, Merck) was purified by crystallisation from chlorobenzene and sublimed. Diethyl (*E*)-2,3-dicyanobutenedioate (10) and 7,7',8,8'-tetracyanoquino-dimethane (16, TCNQ) were purchased from Aldrich and used as purchased.

Reactions of 1-substituted-2,5-dithiobiureas **1a–c** with ethenetetracarbonitrile **(2)**

To a chilled solution of 256 mg (2.0 mmol) of 2 in 10 ml of tetrahydrofuran (THF) a pre-cooled solution of 1a-c in 15 ml of THF was added dropwise which caused a spontaneous change of colour from yellow to pink. The mixture was stirred for 3 h and allowed to warm up to room temperature. The stirring was continued for 48 h with admission of air to complete the reaction. After concentration, the reaction mixture to dryness, the residue sublimed at 80°C under vacuum to remove all unreacted 2. The residue was then separated by PLC (100 mg per plate) using a suitable solvent mixture as aluent (cyclohexane/ethyl acetate 2:1 for the reaction of 2 with 1a and 1b; cyclohexane/ethyl acetate 3:1 for the reaction of 2 with 1c) to give numerous coloured zones. The two most intense of which were removed and extracted. The fastest migrating one contained imidazothiazole derivatives 5a-c, the second zone, which quenched all indicator fluorescence upon exposure to 254 nm UVlight, contained 1,2-bis[2-(substituted amino)-1,3,4-thiadiazol-2yl]hydrazine 6a-c. Extraction of the zones with acetone gave a residue, which was rechromatographed and recrystallised to separate the pure compounds.

2-(*Phenylamino*)*imidazo*[2,1-*b*][1,3,4]*thiadiazole-5,6-dicarbonitrile* (5a): Reddish brown crystals (0.149 g, 56%), m.p. 302–304°C (methanol). ¹H NMR (DMSO-d₆): $\delta = 9.76$ (br, 1H, phenyl-NH), 7.11–7.66 (m, 5H, ArH). ¹³C NMR: $\delta = 156.88$ (C-2), 154.53 (C-4a), 143.33 (ArC), 127.31, 128.82, 129.67 (ArCH), 122.24, 122.26, (C-5, C-6), 118.11, 118.14 (CN). IR (KBr): 3380 (NH), 2215 (CN), 1635 (C=N), 1590 (ArC=C). MS (*m/z*%): 266 (M⁺, 22), 240 (14), 214 (3), 190 (8), 135 (44), 77 (100), 65 (42). C₁₂H₆N₆S (266.28): Calcd; C, 54.13; H, 2.27; N, 31.56; S, 12.04. Found: C, 53.89; H, 2.41; N, 31.78; S, 11.81.

2-(*Benzylamino*)*imidazo*[2,1-*b*][1,3,4]*thiadiazole-5*,6-*dicarbonitrile* (**5b**): Reddish brown crystals (0.148 g, 53%), m.p. 325–327°C (acetonitrile). ¹H NMR (DMSO-d₆): $\delta = 8.85$ (br, 1H, benzyl-NH), 7.06–7.58 (m, 5H, ArH), 4.62 (s, 2H, CH₂). ¹³C NMR: $\delta = 156.66$ (*C*-2), 154.31 (*C*-4a), 122.22, 122.25 (*C*-5, *C*-6), 118.10, 118.13 (*C*N), 52.41 (*C*H₂). IR (KBr): 3365 (NH), 2980 (Ali-CH), 2220 (CN), 1630 (C=N), 1600 (ArC=C). MS (*m/z*,%): 280 (M⁺, 19), 204 (12), 149 (72), 91 (100), 65 (28). C₁₃H₈N₆S (280.31): calcd; C, 55.70; H, 2.88; N, 29.98; S, 11.44. Found: C, 55.93; H, 2.65; N, 30.16; S, 11.67.

2-(*Allylamino*)*imidazo*[2,1-*b*][1,3,4]*thiadiazole-5,6-dicarbonitrile* (**5c**): Reddish brown crystals (0.117 g, 51%), m.p. 271–273°C (ethanol). ¹H NMR (DMSO-d₆): δ = 7.59 (br, 1H, allyl-NH), 5.90–5.93 (m, 1H, allyl-CH=), 5.01–5.18 (m, 2H, allyl-CH₂=), 4.05 (br, 2H, allyl-CH₂N). ¹³C NMR: δ = 156.77 (*C*-2), 155.11 (*C*-4a), 134.21 (allyl-*CH*=), 122.15, 122.18 (*C*-5, *C*-6), 118.08, 118.10 (*C*N), 114.81 (allyl-*CH*=), 43.33 (allyl-*CH*₂N). IR (KBr): 3370 (NH), 2965 (Ali-CH), 2215 (CN), 1635 (C=N). MS (*m*/*z*,%): 230 (M⁺, 36), 204 (6), 154 (44), 99 (82), 41 (100). C₉H₆N₆S (230.25): Calcd; C, 46.95; H, 2.63; N, 36.50; S, 13.93. Found: C, 47.18; H, 2.81; N, 36.29; S, 14.16.

1,2-Bis[5-(phenylamino)-1,3,4-thiadiazol-2-yl]hydrazine (6a): Yellow crystals (0.122 g, 32%), m.p. 240–242°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 9.81 (br, 2H, phenyl-NH), 7.06–7.49 (m, 10H, ArH), 6.61 (br, 2H, hydrazine-NH). ¹³C NMR: δ = 155.76 (*C*-2), 154.33 (*C*-5), 143.12 (ArC), 127.63, 128.81, 129.67 (ArCH). IR (KBr): 3370–3340 (NH), 1620 (C=N), 1595 (ArC=C). MS (m/z,%): 382 (M⁺, 21), 354 (11), 261 (34), 219 (100), 150 (92), 135 (33), 77 (62), 65 (21). C_{16}H_{14}N_8S_2 (382.47): Calcd; C, 50.25; H, 3.69; N, 29.30; S, 16.77. Found: C, 50.44; H, 3.52; N, 29.14; S, 16.96.

1,2-Bis[5-(benzylamino)-1,3,4-thiadiazol-2-yl]hydrazine (6b): Pale yellow crystals (0.148 g, 36%), m.p. 262–264°C (acetonitrile). ¹H NMR (DMSO-d₆): $\delta = 8.87$ (br, 2H, benzyl-NH), 7.10–7.38 (m, 10H, ArH), 6.57 (br, 2H, hydrazine-NH), 4.61 (s, 2H, CH₂). ¹³C NMR: $\delta = 155.48$ (C-2), 154.21 (C-5), 141.63 (ArC), 127.56, 128.82, 129.59 (ArCH), 52.58 (CH₂). IR (KBr): 3385-3360 (NH), 1625 (C=N), 1600 (ArC=C). MS (*m*/z,%): 410 (M⁺, 16), 382 (22), 261 (46), 164 (76), 149 (81), 91 (100), 65 (36). C₁₈H₁₈N₈S₂ (410.52): Calcd; C, 52.66; H, 4.42; N, 27.30; S, 15.62. Found: C, 52.43; H, 4.64; N, 27.53; S, 15.39.

1,2-Bis[5-(allylamino)-1,3,4-thiadiazol-2-yl]hydrazine (6c): Pale yellow crystals (0.105 g, 34%), m.p. 195–197°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 7.61 (br, 2H, allyl-NH), 5.88–5.94 (m, 2H, allyl-CH=), 5.08–5.16 (m, 4H, allyl-CH₂=), 4.08 (br, 4H, allyl-CH₂N),; ¹³C NMR: δ = 155.83 (C-2), 154.71 (C-5), 134.28 (allyl-CH=), 115.08 (allyl-CH₂=), 43.28 (allyl-CH₂N). IR (KBr): 3365– 3345 (NH), 1615 (C=N). MS (*m*/z,%): 310 (M⁺, 23), 282 (14), 269 (27), 211 (16), 131 (89), 114 (46), 99 (62), 41 (100). C₁₀H₁₄N₈S₂ (310.41): Calcd; C, 38.69; H, 4.55; N, 36.10; S, 20.66. Found: C, 38.48; H, 4.71; N, 38.88; S, 20.89.

Reaction of 1-substituted-2,5-dithiobiureas 1a-c with diethyl (E)-2,3dicyanobutenedioate (10): A solution of 444 mg (2.0 mmol) of 10 in 30 ml THF was treated with a) 226 mg (1.0 mmol) of 1a, b) 240 mg (1.0 mmol) of 1b, c) 190 mg (1.0 mmol) of 1c and stirred at 20°C for 48 h (a) and 96 h (b, c), respectively. During which time the colour of the solution changed from yellow to light red. Concentration of the reaction mixture to dryness, the residues was subjected to PLC using cyclohexane/ethyl acetate (5:1) as eluent. Chromatographic separation of the residue gave numerous coloured zones, two of which (with high intensity) were removed and extracted. The fastest move zone contained the thiadiazinylidene cyanoacetate derivatives 11a-c, while the slowest moving zone contained the thiadiazine derivatives 12a-c. Extraction of the zones with acetone and recrystallised gave the pure compounds.

Ethyl (*Z*)-2-[5-amino-2-(phenylamino)-6H-1,3,4-thiadiazin-6-ylidene]-2-cyanoacetate (**11a**): Orange crystals (0.154 g, 49%), m.p. 225– 227°C (ethanol). ¹H NMR (DMSO-d₆): δ = 9.81 (br, 1H, Phenyl-NH), 6.98–7.46 (m, 5H, ArH), 6.37 (br, 2H, NH₂), 4.19 (q, 2H, *J* = 7.15, OCH₂), 1.27 (t, 3H, *J* = 7.15, CH₃). ¹³C NMR: δ = 165.74 (COO), 163.42 (C-5), 161.23 (C-2), 146.65 (C-6), 144.32 (ArC), 127.63, 128.84, 129.61 (ArCH), 118.24 (CN), 115.14 (ethylenic-C), 60.76 (CH₂O), 14.28 (CH₃). IR (KBr): 3360, 3280 (NH₂, NH), 2220 (CN), 1680 (CO), 1610 (C=N), 1580 (ArC=C). MS (*m*/z,%): 315 (M⁺, 41), 299 (14), 204 (28), 180 (11), 150 (21), 135 (12), 77 (100). C₁₄H₁₃N₅O₂S (315.35): Calcd; C, 53.32; H, 4.16; N, 22.21; S, 10.17. Found: C, 53.49; H, 3.98; N, 22.48; S, 10.36.

Ethyl (*Z*)- 2-[5-amino-2-(benzylamino)-6H-1,3,4-thiadiazin-6-ylidene]-2-cyanoacetate (**11b**): Orange crystals (0.151 g, 46%), m.p. 258– 260°C (methanol). ¹H NMR (DMSO-d₆): $\delta = 8.89$ (br, 1H, benzyl-NH), 7.05–7.44 (m, 5H, ArH), 6.34 (br, 2H, NH₂), 4.62 (s, 2H, PhCH₂), 4.16 (q, 2H, J = 7.10, OCH₂), 1.25 (t, 3H, J = 7.10, CH₃). ¹³C NMR: $\delta = 165.56$ (COO), 163.12 (C-5), 161.44 (C-2), 146.65 (C-6), 143.61 (ArC), 127.48, 128.68, 129.14 (ArCH), 118.19 (CN), 114.99 (ethylenic-C), 60.79 (CH₂O), 52.64 (Ph-CH₂), 14.26 (CH₃). IR (KBr): 3350, 3225 (NH₂, NH), 2225 (CN), 1685 (CO), 1620 (C=N), 1590 (ArC=C). MS (m/z,%): 329 (M⁺, 26), 313 (8), 303 (6), 218 (15), 180 (16), 149 (33), 91 (100), 77 (84). C₁₅H₁₅N₅O₂S (329.38): Calcd; C, 54.70; H, 4.59; N, 21.26; S, 9.74. Found: C, 54.45; H, 4.80; N, 21.54; S, 10.02.

Ethyl(*Z*)-2-[5-amino-2-(allylamino)-6H-1,3,4-thiadiazin-6-ylidene]-2-cyanoacetate (11c): Orange crystals (0.126 g, 45%), m.p. 196– 198°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 7.61 (br, 1H, allyl-NH), 6.40 (br, 2H, NH₂), 5.88–5.91 (m, 1H, allyl-CH=), 5.16–5.22 (m, 2H, allyl-CH₂=), 4.18 (q, 2H, *J* = 7.06, OCH₂), 4.05 (br, 2H, allyl-CH₂N), 1.26 (t, 3H, *J* = 7.06, CH₃). ¹³C NMR: δ = 165.32 (COO), 163.22 (*C*-2), 161.28 (*C*-2), 146.55 (*C*-6), 134.29 (allyl-*CH*=), 117.96 (CN), 115.88 (allyl-CH₂=), 114.15 (ethylenic-C), 60.93 (*C*H₂O), 43.12 (allyl-CH₂N), 14.26 (CH₃), IR (KBr): 3370, 3290 (NH₂, NH), 2220 (CN), 1680 (CO), 1615 (C=N). MS (*m*/*z*,%): 279 (M⁺, 16), 263 (9), 203 (13), 180 (23), 99 (91), 41 (100). C₁₁H₁₃N₅O₂S (279.32): Calcd; C, 47.30; H, 4.69; N, 25.07; S, 11.48. Found: C, 47.49; H, 4.52; N, 24.87; S, 11.27.

Ethyl 5-amino-6-cyano-2-(phenylamino)-6(H)-1,2,4-thiadiazine-6-carboxylate (12a): Reddish brown crystals (0.124 g, 41%), m.p. 205–207°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 9.84 (br, 1H,

phenyl-NH), 7.09–7.46 (m, 5H, ArH), 6.39 (br, 2H, NH₂), 4.16 (q, 2H, J = 7.03, OCH₂), 1.25 (t, 3H, J = 7.03, CH₃). ¹³C NMR: $\delta = 166.14$ (COO), 159.11 (C-5), 158.77 (C-2), 144.53 (ArC), 127.38, 128.82, 129.69 (ArCH), 117.97 (CN), 60.76 (CH₂O), 45.18 (C-6), 14.17 (CH₃). IR (KBr): 3350, 3260 (NH₂, NH), 2210 (CN), 1685 (CO), 1610 (C=N). MS (*m*/*z*,%): 303 (M⁺, 34), 287 (8), 259 (11), 229 (9), 168 (26), 135 (74), 77 (100). C₁₃H₁₃N₅O₂S (303.34): Calcd; C, 51.47; H, 4.32; N, 23.09; S, 10.57. Found: C, 51.69; H, 4.11; N, 22.85; S, 10.81.

Ethyl 5-amino-6-cyano-2-(benzylamino)-6(H)-1,2,4-thiadiazine-6-carboxylate (12b): Reddish brown crystals (0.124 g, 39%), m.p. 237–239°C (acetonitrile). ¹H NMR (DMSO-d₆): $\delta = 8.91$ (br, 1H, benzyl-NH), 7.22–7.39 (m, 5H, ArH), 6.43 (br, 2H, NH₂), 4.64 (s, 2H, PhCH₂), 4.18 (q, 2H, J = 7.12, OCH₂), 1.23 (t, 3H, J = 7.12, CH₃). ¹³C NMR: $\delta = 165.76$ (COO), 159.26 (C-5), 158.69 (C-2), 127.82, 127.96, 128.84 (ArCH), 118.14 (CN), 141.12 (ArC), 60.71 (CH₂O), 52.61 (Ph-CH₂), 44.96 (C-6), 14.22 (CH₃). IR (KBr): 3365, 3280 (NH₂, NH), 2220 (CN), 1690 (CO), 1620 (C=N). MS (*m*/z,%): 317 (M⁺, 29), 290 (6), 243 (16), 168 (33), 149 (52), 91 (100), 77 (62). C₁₄H₁₅N₅O₂S (317.37): Calcd; C, 52.98; H, 4.76; N, 22.07; S, 10.10. Found: C, 53.22; H, 4.59; N, 21.86; S, 10.36.

Ethyl 5-amino-6-cyano-2-(allylamino)-6(H)-1,2,4-thiadiazine-6carboxylate (12c): Reddish brown crystals (0.96 g, 36%), m.p. 181– 183°C (ethanol). ¹H NMR (DMSO-d₆): δ = 7.55 (br, 1H, allyl-NH), 6.35 (br, 2H, NH₂), 5.90–5.94 (m, 1H, allyl-CH=), 5.18–5.26 (m, 2H, allyl-CH₂=), 4.18 (q, 2H, *J* = 6.97, OCH₂), 4.09 (br, 2H, allyl-NH₂), 1.22 (t, 3H, *J* = 6.97, CH₃). ¹³C NMR: δ = 166.12 (COO), 159.67 (*C*-5), 158.93 (*C*-2), 134.22 (allyl-CH=), 118.16 (CN), 116.06 (allyl-CH₂=), 60.63 (CH₂O), 45.03 (*C*-6), 42.88 (allyl-CH₂N), 14.19 (CH₃). IR (KBr): 3360, 3270 (NH₂, NH), 2960 (Ali-H), 2220 (CN), 1685 (CO), 1610 (C=N). MS (*m*/z,%): 267 (M⁺, 26), 193 (19), 168 (33), 165 (5), 99 (67), 41 (100). C₁₀H₁₃N₅O₂S (267.31): Calcd; C, 44.93; H, 4.90; N, 26.20; S, 12.00. Found: C, 45.21; H, 4.72; N, 25.97; S, 11.97.

Reaction of 1-substituted-2,5-dithiobiureas 1a-c with 7,7',8,8'-tetracvanoquinodimethane (16): To a solution of 16 (416 mg, 2.0 mmol) in dry pyridine (15 ml) a solution of **1a-c** (1.0 mmol each) in (5 ml) of dry pyridine was added dropwise over 5 minutes at room temperature with stirring. The mixture was warmed gently to 50-60°C and kept at this temperature with stirring and admission of air for 3 h, then warmed to maximum 100°C for few minutes and concentrated to dryness at 50°C. The residue was taken up several times with cold ethanol (15 ml) and the slurry was concentrated again to remove any residual pyridine. This operation was repeated four times. The residue was dissolved in (5 ml) acetone. This solution in each case was applied to five PLC plates and developed with toluene/ethyl acetate (3:1) for the run with 1a, toluene/ethyl acetate (4:1) for the run with 1b and 1c. Numerous zones were observed, two of which were extracted. The fastest migrating zone contained compound 17, the second zone contained compound 18. Extraction of the zones with acetone gave a residue, which was rechromatographed with the same eluent to enhance separation. Recrystallisation from suitable solvents afforded compounds 17a-c and 18a-c.

{5-*Amino*-6-*cyano*-2-(*phenylamino*)*spiro*[1,3,4]*thiadiazolo*[3, 2-*a*]*pyrimidine*-7,1'-*cyclohexa*[2,5]*dien*-4'-*ylidene*}*malononitrile* (17a): Blue crystals (0.234 g, 59%), m.p. 246–248°C (methanol). ¹H NMR (DMSO-d₆): $\delta = 9.77$ (br, 1H, phenyl-NH), 7.06–7.46 (m, 5H, ArH), 6.74 (br, 2H, NH₂), 6.32, 6.61 (dd, 4H, cyclohexadiene-H). ¹³C NMR: $\delta = 166.32$ (C-4'), 161.12 (C-5), 157.12 (C-8a), 155.72 (C-2), 144.61 (ArC), 127.62, 128.83, 129.64 (ArCH), 124.71, 126.82 (cyclohexadiene-CH), 118.66 (CN), 73.84 (malononitrile-C), 60.62 (C-6), 45.77 (q-C-7,1'). IR (KBr): 3410, 3310 (NH₂, NH), 2225, 2220 (CN), 1630 (C=N), 1595 (ArC=C). MS (*m*/z,%): 396 (M⁺, 31), 330 (11), 190 (19), 135 (51), 77 (100). C₂₀H₁₂N₈S (396.43): Calcd; C, 60.59; H, 3.05; N, 28.27; S, 8.09. Found: C, 60.37; H, 2.94; N, 28.51; S, 8.32.

{5-Amino-6-cyano-2-(benzylamino)spiro[1,3,4]thiadiazolo[3, 2-a]pyrimidine-7,1'-cyclohexa[2,5]dien-4'-ylidene}malononitrile (17b): Blue crystals (0.226 g, 55%), m.p. 276–278°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 8.91 (br, 1H, benzyl-NH), 6.98–7.35 (m, 5H, ArH), 6.70 (br, 2H, NH₂), 6.30, 6.58 (dd, 4H, cyclohexadiene-H), 4.62 (s, 2H, CH₂). ¹³C NMR: δ = 166.22 (C-4'), 160.91 (C-5), 156.88 (C-8a), 155.57 (C-2), 141.77 (ArC), 127.83, 127.91, 128.68 (ArCH), 124.79, 126.77 (cyclohexadiene-CH), 118.61 (CN), 73.69 (malononitrile-C), 60.45 (C-6), 52.61 (CH₂), 45.53 (q-C-7,1'). IR (KBr): 3395, 3315 (NH₂, NH), 2220, 2215 (CN), 1625 (C=N), 1595 (ArC=C). MS (*m*/z,%): 410 (M⁺, 23), 344 (9), 204 (22), 149 (66), 91 (100), 77 (58). C₂₁H₁₄N₈S (410.45): Calcd; C, 61.45; H, 3.44; N, 27.30; S, 7.81. Found: C, 61.68; H, 3.65; N, 27.57; S, 7.55.

{5-Amino-6-cyano-2-(allylamino)spiro[1,3,4]thiadiazolo[3,2-a] pyrimidine-7,1'-cyclohexa[2,5]dien-4'-ylidene}malononitrile (17c): Blue crystals (0.105 g, 57%), m.p. 216–218°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 7.61 (br, 1H, ally-NH), 6.65 (br, 2H, NH₂), 6.33, 6.61 (dd, 4H, cyclohexadiene-H), 5.88–5.92 (m, 1H, allyl-CH=), 5.20–5.24 (m, 2H, allyl-CH₂=), 4.05 (br, 2H, allyl-CH₂N). ¹³C NMR: δ = 165.98 (C-4'), 160.87 (C-5), 156.75 (C-8a), 155.57 (C-2), 134.31 (allyl-CH=), 124.72, 126.64 (cyclohexadiene-CH), 118.61 (CN), 116.14 (allyl-CH₂=), 73.75 (malononitrile-C), 60.51 (C-6), 45.48 (q-C-7,1'), 42.85 (allyl-CH₂N). IR: 3400, 3320 (NH₂, NH), 2225, 2220 (CN), 1630 (C=N). MS (*m*/*z*,%): 360 (M⁺, 22), 294 (9), 163 (26), 99 (57), 41 (100). C₁₇H₁₂N₈S (360.40): Calcd; C, 56.65; H, 3.36; N, 31.09; S, 8.90. Found: C, 56.81; H, 3.22; N, 30.89; S, 9.14.

1,4-Diamino-5-cyano-9-(dicyanomethylene)-N-phenyl-2, 3-diazospiro[5.5]undeca-1,4,7,10-tetraene-9-thioamide (18a): Violet crystals (0.111 g, 28%), m.p. 289–291°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 9.84 (br, 1H, phenyl-NH), 7.09–7.70 (m, 5H, ArH), 6.62 (br, 2H, NH₂), 6.49 (br, 2H, NH₂), 6.29, 6.58 (dd, 4H, cyclohexadiene-H). ¹³C NMR: δ = 180.19 (C=S), 166.41 (C-9), 160.83 (C-4), 156.67 (C-1), 141.18 (ArC), 127.86, 128.88, 129.51 (ArCH), 125.13, 126.64 (cyclohexadiene-CH), 117.97, 118.31 (CN), 74.26 (malononitrile-C), 60.71 (C-5), 46.21 (q-C-6). IR (KBr): 385, 3270 (NH₂, NH), 2225, 2215 (CN), 1610 (C=N), 1570 (NH def. and C-N str.), 1358, 998 (C=S, C-N). MS (m/z,%): 398 (M⁺, 27), 263 (35), 207 (36), 197 (21), 169 (15), 150 (42), 135 (57), 77 (100), 66 (46), 57 (37). C₂₀H₁₄N₈S (398.44): Calcd; C, 60.29; H, 3.54; N, 28.12; S, 8.05. Found: C, 60.51; H, 3.33; N, 27.89; S, 8.31.

1, 4-Diamino-5-cyano-9-(dicyanomethylene)-N-benzyl-2, 3-diazospiro[5.5]undeca-1,4,7,10-tetraene-9-thioamide (18b): Violet crystals (0.136 g, 33%), m.p. 324–326°C (acetonitrile). ¹H NMR (DMSO-d₆): δ = 8.91 (br, 1H, benzyl-NH), 7.16–7.53 (m, 5H, ArH), 6.54 (br, 2H, NH₂), 6.46 (br, 2H, NH₂), 6.25, 6.55 (dd, 4H, cyclohexadiene-H), 4.62 (CH₂). ¹³C NMR: δ = 180.46 (C=S), 166.72 (C-9), 161.12 (C-4), 156.41 (C-1), 141.66 (ArC), 127.84, 127.96, 128.62 (ArCH), 125.06, 126.51 (cyclohexadiene-CH), 118.05, 118.27 (CN), 74.28 (malononitrile-C), 60.66 (C-5), 52.56 (CH₂), 45.96 (q-C-6). IR (KBr): 3370, 3280 (NH₂, NH), 2225, 2215 (CN), 1620 (C=N), 1565 (NH def. and C-N str.), 1355, 1005 (C=S, C-N). MS (*m*/z,%): 412 (M⁺, 18), 263 (14), 207 (9), 169 (22), 164 (26), 149 (77), 91 (100), 77 (66), 66 (51), 57 (24). C₂₁H₁₆N₈S (412.47): Calcd; C, 61.15; H, 3.91; N, 27.17; S, 7.77. Found: C, 60.92; H, 4.08; N, 27.41; S, 7.59.

1,4-Diamino-5-cyano-9-(dicyanomethylene)-N-allyl-2,3-diazospiro[5.5]undeca-1,4,7,10-tetraene-9-thioamide (**18c**): Violet crystals (0.101 g, 28%), m.p. 261–263°C (acetonitrile). ¹H NMR (DMSO-d₆): $\delta = 7.84$ (br, 1H, allyl-NH), 6.57 (br, 2H, NH₂), 6.48 (br, 2H, NH₂), 6.28, 6.60 (dd, 4H, cyclohexadiene-H), 5.92–5.96 (m, 1H, allyl-CH=), 5.18–5.26 (m, 2H, allyl-CH₂=), 4.10 (br, 2H, allyl-CH₂N). ¹³C NMR: $\delta = 180.52$ (*C=S*), 166.12 (*C*-9), 160.88 (*C*-4), 156.36 (*C*-1), 134.38 (allyl-*CH*=), 125.11, 126.49 (cyclohexadiene-*C*H), 118.06, 118.26 (*C*N), 115.96 (allyl-*CH*₂=), 74.36 (malononitrile-*C*), 60.58 (*C*-5), 46.08 (q-*C*-6), 42.78 (allyl-*CH*₂N). IR (KBr): 3345, 3270 (NH₂, NH), 2220, 2210 (CN), 1560 (NH def. and C-N str.), 1355, 1015 (C=S, C–N). MS (*m*/*z*,%): 362 (M⁺, 14), 263 (26), 207 (14), 197 (16), 181 (8), 169 (19), 99 (46), 66 (58), 57 (45), 41 (100). C₁₇H₁₄N₈S (362.41): Calcd; C, 56.34; H, 3.89; N, 30.92; S, 8.85. Found: C, 56.11; H, 4.11; N, 31.14; S, 8.63.

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References

- 1 A.J. Fataidi, Synthesis, 1986, 249.
- 2 P. Bruni and G. Tosi, Gazz Chim. Ital. 1997, 127, 435.
- 3 G. Bocelli, C.R. Cardellini, G. Demeo, A. Ricci, C. Rizzoli and G. Tosi, J. Crystallogr. Spectrosc. Res., 1990, 20, 561.
- 4 A.J. Fatiadi, Synthesis, 1987, 749.
- 5 A.J. Fatiadi, Synthesis, 1987, 749.
- 6 D. Döpp, S. Jüschke and G. Henkel, Z. Naturforsch, 2002, 57b, 460.
- 7 D. Döpp, A.A. Hassan, A.E. Mourad, A.M. Nour El Din, K. Argermund, C. Krüger, C.W. Lehmann and J. Rust, *Tetrahedron*, 2003, **59**, 5073.
- A. Hassan, K.M. El-Shaieb and D. Döpp, Arkivoc, 2005, x, 139.
- 9 A.A. Hassan, Phosphorus, Sulfur, Silicon 1995, 101, 189.
- 10 A.A. Hassan, Y.R. Ibrahim, E.H. El-Tamany, A.A. Samida and A.E. Mourad, *Phosphorus, Sulfur, Silicon*, 1995, 167.

- 11 A.A. Hassan, N.K. Mohamed, A.A. Aly and A.E. Mourad, Monatsh. Chem., 1997, 128, 61.
- 12 A.A. Hassan, N.K. Mohamed, A.M. Shawky and D. Döpp, Arkivoc, 2003, i, 118.
- 13 A.A. Hassan, A.E. Mourad, K.M. El-Shaieb and A.H. Abou-Zied,
- Z. Naturforsch, 2004, **59b**, 910. A.A. Hassan, A.E. Mourad, K.M. El-Shaieb, R.M. Shaker and D. Döpp, *Heteroatom Chem.*, 2005, **16**, 12. 14
- 15 M.A.-M. Gomaa, A.A. Hassan and H.S. Shehatta, Heteroatom Chem., 2006, 17, 261.
- A.A. Hassan, A.E. Mourad and A.H. Abou-Zied, J. Heterocyclic Chem., 16 2007, 44, 1171.
- 17 B.B. Kanl, W.S. Durfee and G.T. Yee, J. Am. Chem. Soc., 1999, 121, 6862
- 18 Y. Yamada, H. Yasuda and M. Kasai, Heterocycles, 1999, 51, 2453. 19
- Y. Yamada and H. Yasuda, J. Heterocyclic Chem., 1998, 35, 1389. Y. Yamada, H. Yasuda, Y. Yoshihara and K. Yoshizawa, J. Heterocyclic 20
- Chem., 1999, 36, 1327.
- 21 R.C. Wheland and E.L. Martin, J. Org. Chem., 1975, 40, 3101.

- 22 N. Martin and C. Seoane, Organic Conductive Molecules and Polymers, N.S. Nalwa, Wiley, New York, 1996.
 R.M. Metzger, R.R. Shumaker, M.P. Cara, R.K. Laidlow, L.A. Panetta
- and E. Torres, Langmuir, 1988, 4, 198.
- 24 G.J. Ashwell, E. Je. Downay, A.P. Kuczynski, M. Szablewski, I.M. Sandy, M.R. Bryce, A.M. Girainger and M. Hasan, J. Chem. Soc., Faraday Trans., 1990, 86, 1117.
 J.S. Miller, J.C. Calabrese, R.L. Harlow, D.A. Dixan, J.H. Zhang, W.M. Rieff, S. Chittipeddi, M.A. Selower and A.J. Epskin, J. Am. Chem.
- Soc., 1990, 112, 5496.
- 26 N.A. Zaidi, M.R. Bryce and G.H. Cross, Tetrahedron Lett., 2000, 41, 4645.
- A.A. Aly and K.M. El-Shaieb, Teterahedron, 2004, 60, 3797. 27 28 H.-O. Kalinowski, S. Berger and S. Brann, ¹³C NMR Specyroscopic, Thieme, Stuttgart, 1984, pp 121.
- K. Gewald and R. Schnidler, J. Prakt. Chem., 1990, 332, 223.
 P.V. Indukumari, C.P. Josshva and V.P. Rajan, Indian J. Chem., 1981, 20B, 201 384
- 31 W.H. Altland and A.P. Graham, J. Heterocyclic Chem., 1978, 15, 377.